

Chemistry

Date	Exam	Unit(s)	
		Teacher 1 (5L)	Teacher 2 (4L)
4/9/23		<b>Thermodynamics</b> <ul style="list-style-type: none"> <li>Enthalpy changes in ionic compounds</li> <li>Born-Haber cycles</li> <li>Perfect ionic model</li> <li>Enthalpy of solution</li> <li>Feasible reactions</li> <li>Entropy</li> <li>Feasibility of a reaction</li> <li>Graphical calculations</li> </ul> <b>Equilibrium constant Kp</b> <ul style="list-style-type: none"> <li>Mole fractions</li> <li>Partial pressure</li> <li>Kp</li> </ul>	<b>Kinetics</b> <ul style="list-style-type: none"> <li>Methods of determining rate of reaction</li> </ul> Required Practical 7a. An 'iodine clock' experiment – initial rate method. Required Practical 7b. Measuring the rate of reaction by a continuous monitoring method. <ul style="list-style-type: none"> <li>Rate monitoring</li> <li>Order linked to mechanisms</li> <li>The rate equation</li> <li>The Arrhenius equation</li> </ul> <b>Optical isomerism</b> <ul style="list-style-type: none"> <li>Enantiomers</li> <li>Optical activity</li> </ul>
11/9/23			
18/9/23			
25/9/23			
2/10/23			
09/10/23			
16/10/23			
30/10/23	PPE1	<b>PPE1</b> <b>Transition Metals</b> <ul style="list-style-type: none"> <li>General properties</li> <li>Complex formation</li> <li>Ligand substitution</li> <li>Shapes of complex ions</li> <li>Formation of coloured ions</li> <li>Ions in aqueous solutions</li> </ul> <b>Transition Metals: Variable oxidation states</b> <ul style="list-style-type: none"> <li>Oxidation states</li> <li>Redox titrations</li> <li>Catalytic activity</li> </ul> Required Practical 11. Identify transition metal ions in aqueous solution.	<b>PPE1</b> <b>The carbonyl group</b> <ul style="list-style-type: none"> <li>Aldehydes &amp; ketones</li> <li>Nucleophilic addition reactions</li> <li>Carboxylic acids &amp; derivatives</li> <li>Esters</li> </ul> Required Practical 10b. Preparation of a pure organic liquid – ethyl ethanoate. <ul style="list-style-type: none"> <li>Fats &amp; oils</li> <li>Acylation – acid chlorides &amp; anhydrides</li> <li>Nucleophilic addition-elimination</li> </ul> Required Practical 10a. Preparation of an organic solid and a test of its purity – aspirin.
6/11/23	PPE1		
13/11/23			
20/11/23			
27/11/23			
4/12/23			
11/12/23			
1/1/24		<b>Electrode potentials &amp; cells</b> <ul style="list-style-type: none"> <li>Redox equilibria</li> <li>Cell conventions &amp; EMF</li> <li>Redox reactions &amp; feasibility</li> <li>Conventional cell representation</li> </ul> Required Practical 8. Measuring the EMF of an electrochemical cell. <ul style="list-style-type: none"> <li>Commercial applications</li> </ul> <b>PPE2</b>	<b>Aromatic chemistry</b> <ul style="list-style-type: none"> <li>Structure of benzene</li> <li>Addition &amp; substitution reactions</li> <li>Electrophilic substitution</li> </ul> <b>Amines</b> <ul style="list-style-type: none"> <li>Structure &amp; naming</li> <li>Preparation of primary &amp; aromatic amines</li> <li>Basic properties</li> <li>Nucleophilic substitution</li> </ul> <b>PPE2</b>
8/1/24			
15/1/24			
22/1/24			
29/1/24			
5/2/24	PPE2		
12/2/24	PPE2		
26/2/24		<b>Acids &amp; Bases</b> <ul style="list-style-type: none"> <li>Bronsted Lowry theory</li> <li>pH</li> <li>Ionic product of water</li> <li>pH of weak acids</li> <li>Dilutions &amp; neutralisations</li> <li>Titration curves</li> <li>Buffers</li> </ul> Required Practical 9. Investigate how pH changes when a weak acid reacts with a strong base.	<b>Polymers</b> <ul style="list-style-type: none"> <li>Addition polymerisation</li> <li>Condensation polymerisation</li> <li>Biodegradability of polymers</li> </ul> <b>Biological molecules</b> <ul style="list-style-type: none"> <li>Amino acids</li> <li>Proteins &amp; enzyme action</li> <li>DNA</li> </ul> <b>Structure determination &amp; analysis</b> <ul style="list-style-type: none"> <li>Chromatography</li> </ul> Required Practical 12. Separation of a species by thin-layer chromatography. <ul style="list-style-type: none"> <li>NMR spectroscopy</li> <li>Functional group tests</li> </ul>
4/3/24			
11/3/24			
18/3/24			
25/3/24			
15/4/24		<b>Properties of Period 3 elements &amp; their oxides</b> <ul style="list-style-type: none"> <li>Elements of Period 3</li> <li>Oxides</li> <li>Trends in melting points</li> <li>Structure of the acids &amp; anions</li> </ul> <b>Revision &amp; exam preparation</b>	<b>Organic Synthesis</b> <ul style="list-style-type: none"> <li>Revision &amp; summary of all organic reactions</li> </ul> <b>Revision &amp; exam preparation</b>
22/4/24			
29/4/24			
6/5/24			

Chemistry

13/5/24			
20/5/24		A-level exam season provisional start date	A-level exam season provisional start date
<b>3/6/24</b>		<b><i>A-Level/Vocational Examinations</i></b>	<b><i>A-Level/Vocational Examinations</i></b>
10/6/24			
17/6/24			
24/6/24			
1/7/24			
8/7/24			
15/7/24			